

## Comfa Study of A Series of N<sub>1</sub>-(Substituted)Aryl-2-(Substituted)-5,7-Dimethyl-pyrido[2,3-D]Pyrimidin-4(3*h*)-One – Histamine H<sub>1</sub>-Receptor Antagonists

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**ABSTRACT:** In continuation of our efforts to develop a potent and selective peripheral H<sub>1</sub> receptor antagonist for the treatment of allergic rhinitis or urticaria, earlier, we reported potent H<sub>1</sub>-receptor antagonistic activity in a series of N<sub>1</sub>-(substituted)aryl-2-(substituted)-5,7-dimethyl pyrido[2,3-*d*]pyrimidin-4(3*H*)-one. Three-dimensional quantitative structure-activity relationship (3D-QSAR) studies were performed on H<sub>1</sub> receptor antagonists based on Comparative Molecular Field Analysis (CoMFA) method. Significant correlation coefficients,  $r^2 = 0.894$  and  $q^2 = 0.491$  were obtained and the generated model was externally validated using test sets with  $r^2$  predicted 0.624. The CoMFA model can be used to design novel, potent and selective H<sub>1</sub> receptor antagonists prior to their synthesis.

**KEYWORDS:** Pyridopyrimidines; • H<sub>1</sub> receptor antagonist; • 3D-QSAR; • CoMFA; • Tripose.

### Background

Histamine is a neurotransmitter released in a variety of allergic conditions such as seasonal rhinitis (hay fever), urticaria (rash), pruritis, and insect stings as well as a reaction to certain drugs such as penicillin or aspirin. Histamine mediates its activity by interaction at H<sub>1</sub> receptors. Agents that selectively block these receptors are therefore of benefit in histamine-related allergic responses. Although very effective, the first-generation antihistamines which appeared in clinical practice during the 1940s suffered from two major drawbacks. First, they were highly lipid soluble and readily penetrate the blood-brain barrier (BBB) so cause marked sedation and second their

nonselectivity for peripheral H<sub>1</sub> receptors at the given doses, so cause anticholinergic adverse effects, resulting in performance or cognitive function impairment and therapy non-adherence<sup>1-7</sup>.

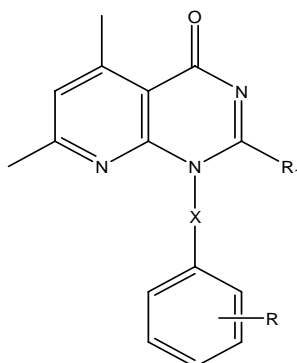
The second and third-generation antihistamines known as the "nonsedating antihistamines" were developed to minimize these side effects. Chemical structures of these drugs often deviate from the basic structure of classical antihistamines but they fulfill specific lipophilicity criteria which prevent them to cross BBB. Their activity is very sensitive to the precise stereochemistry and specific molecular interactions between drug and receptor. In continuation of our efforts to develop novel third generation H<sub>1</sub>-receptor antagonist, we designed, synthesized and screened a series of N<sub>1</sub>-(substituted)aryl-2-(substituted)-5,7-dimethylpyrido[2,3-*d*]pyrimidin-4(3*H*)-one<sup>8</sup>. The structures of these compounds and their biological activities are listed in Table 1.

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**Table 1** Molecular structures of pyridopyrimidine derivatives with their respective log pA<sub>2</sub> values



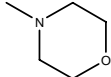
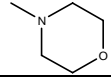
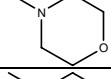
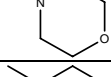
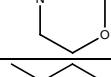
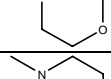
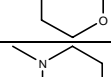
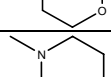
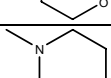
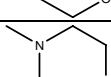
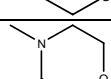
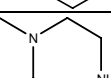
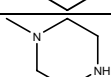
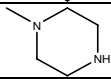
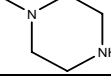
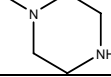
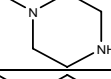
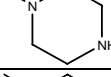
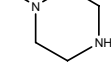
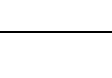
| No. | X               | R                         | R <sub>1</sub>   | log pA <sub>2</sub> |
|-----|-----------------|---------------------------|--|---------------------|
| 1*  | -               | H                         |    | 0.9148              |
| 2   | CH <sub>2</sub> | H                         |    | 0.9375              |
| 3   | -               | m-CH <sub>3</sub>         |    | 0.9079              |
| 4   | -               | p-CH <sub>3</sub>         |    | 0.8779              |
| 5*  | -               | m-OCH <sub>3</sub>        |    | 0.9498              |
| 6   | -               | p-OCH <sub>3</sub>        |    | 0.9164              |
| 7   | -               | p-F                       |    | 0.9434              |
| 8   | -               | m-Cl                      |    | 0.9493              |
| 9   | -               | p-Cl                      |   | 0.9253              |
| 10  | -               | p-Br                      |  | 0.9703              |
| 11  | -               | 2,3- di(CH <sub>3</sub> ) |  | 0.941               |
| 12* | -               | 2,4- di(CH <sub>3</sub> ) |  | 0.939               |
| 13* | -               | H                         |  | 0.875               |
| 14  | CH <sub>2</sub> | H                         |  | 0.8633              |
| 15  | -               | m-CH <sub>3</sub>         |  | 0.9493              |
| 16* | -               | p-CH <sub>3</sub>         |  | 0.941               |
| 17  | -               | m-OCH <sub>3</sub>        |  | 0.9508              |
| 18* | -               | p-OCH <sub>3</sub>        |  | 0.939               |
| 19* | -               | p-F                       |  | 0.9532              |
| 20  | -               | m-Cl                      |  | 0.9405              |

Table 1 Contd...

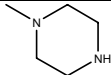
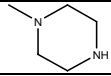
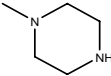
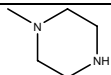
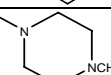
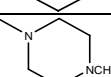
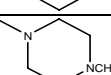
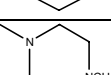
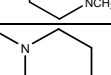
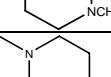
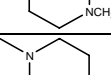
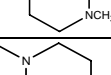
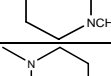
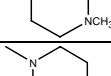
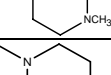
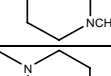
| No. | X               | R                         | R <sub>1</sub>   | log pA <sub>2</sub> |
|-----|-----------------|---------------------------|--|---------------------|
| 21  | -               | p-Cl                      |    | 0.94                |
| 22* | -               | p-Br                      |    | 0.9774              |
| 23  | -               | 2,3- di(CH <sub>3</sub> ) |    | 0.9079              |
| 24* | -               | 2,4- di(CH <sub>3</sub> ) |    | 0.8774              |
| 25  | -               | H                         |    | 0.918               |
| 26  | CH <sub>2</sub> | H                         |    | 0.9159              |
| 27* | -               | m- CH <sub>3</sub>        |    | 0.9232              |
| 28  | -               | p- CH <sub>3</sub>        |    | 0.9132              |
| 29* | -               | m-OCH <sub>3</sub>        |    | 0.9756              |
| 30  | -               | p-OCH <sub>3</sub>        |   | 0.9426              |
| 31* | -               | p-F                       |  | 0.9604              |
| 32  | -               | m-Cl                      |  | 0.9561              |
| 33  | -               | p-Cl                      |  | 0.9365              |
| 34  | -               | p-Br                      |  | 0.9777              |
| 35* | -               | 2,3- di(CH <sub>3</sub> ) |  | 0.9164              |
| 36* | -               | 2,4- di(CH <sub>3</sub> ) |  | 0.9294              |
| 37  | -               | H                         | -N(CH <sub>3</sub> ) <sub>2</sub>  | 0.8677              |
| 38* | CH <sub>2</sub> | H                         | -N(CH <sub>3</sub> ) <sub>2</sub>  | 0.9063              |
| 39  | -               | m- CH <sub>3</sub>        | -N(CH <sub>3</sub> ) <sub>2</sub>  | 0.9077              |
| 40* | -               | p- CH <sub>3</sub>        | -N(CH <sub>3</sub> ) <sub>2</sub>  | 0.8776              |
| 41  | -               | m-OCH <sub>3</sub>        | -N(CH <sub>3</sub> ) <sub>2</sub>  | 0.9263              |
| 42  | -               | p-OCH <sub>3</sub>        | -N(CH <sub>3</sub> ) <sub>2</sub>  | 0.9253              |
| 43* | -               | p-F                       | -N(CH <sub>3</sub> ) <sub>2</sub>  | 0.9518              |
| 44* | -               | m-Cl                      | -N(CH <sub>3</sub> ) <sub>2</sub>  | 0.8621              |
| 45* | -               | p-Cl                      | -N(CH <sub>3</sub> ) <sub>2</sub>  | 0.9222              |
| 46  | -               | p-Br                      | -N(CH <sub>3</sub> ) <sub>2</sub>  | 0.9561              |
| 47  | -               | 2,3- di(CH <sub>3</sub> ) | -N(CH <sub>3</sub> ) <sub>2</sub>  | 0.9395              |
| 48  | -               | 2,4- di(CH <sub>3</sub> ) | -N(CH <sub>3</sub> ) <sub>2</sub>  | 0.9289              |
| 49* | -               | H                         | -N(C <sub>2</sub> H <sub>5</sub> ) <sub>2</sub>                                      | 0.877               |
| 50  | CH <sub>2</sub> | H                         | -N(C <sub>2</sub> H <sub>5</sub> ) <sub>2</sub>                                      | 0.9079              |
| 51  | -               | m- CH <sub>3</sub>        | -N(C <sub>2</sub> H <sub>5</sub> ) <sub>2</sub>                                      | 0.9132              |

Table 1 Contd...

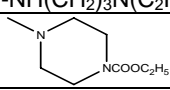
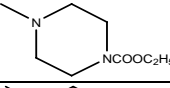
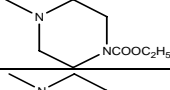
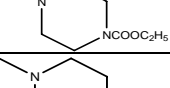
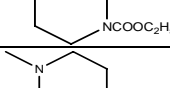
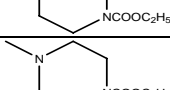
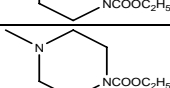
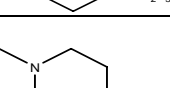
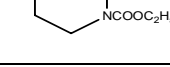
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|-----|-----------------|---------------------------|--|---------------------|
| 52* | -               | p- CH <sub>3</sub>        | -N(C <sub>2</sub> H <sub>5</sub> ) <sub>2</sub>                                      | 0.939               |
| 53  | -               | m-OCH <sub>3</sub>        | -N(C <sub>2</sub> H <sub>5</sub> ) <sub>2</sub>                                      | 0.9459              |
| 54  | -               | p-OCH <sub>3</sub>        | -N(C <sub>2</sub> H <sub>5</sub> ) <sub>2</sub>                                      | 0.9283              |
| 55* | -               | p-F                       | -N(C <sub>2</sub> H <sub>5</sub> ) <sub>2</sub>                                      | 0.9508              |
| 56  | -               | m-Cl                      | -N(C <sub>2</sub> H <sub>5</sub> ) <sub>2</sub>                                      | 0.9561              |
| 57* | -               | p-Cl                      | -N(C <sub>2</sub> H <sub>5</sub> ) <sub>2</sub>                                      | 0.9148              |
| 58  | -               | p-Br                      | -N(C <sub>2</sub> H <sub>5</sub> ) <sub>2</sub>                                      | 0.9774              |
| 59* | -               | 2,3- di(CH <sub>3</sub> ) | -N(C <sub>2</sub> H <sub>5</sub> ) <sub>2</sub>                                      | 0.94                |
| 60  | -               | 2,4- di(CH <sub>3</sub> ) | -N(C <sub>2</sub> H <sub>5</sub> ) <sub>2</sub>                                      | 0.9395              |
| 61* | -               | H                         | -NH(CH <sub>2</sub> ) <sub>2</sub> N(C <sub>2</sub> H <sub>5</sub> ) <sub>2</sub>    | 0.9763              |
| 62  | CH <sub>2</sub> | H                         | -NH(CH <sub>2</sub> ) <sub>2</sub> N(C <sub>2</sub> H <sub>5</sub> ) <sub>2</sub>    | 0.9881              |
| 63  | -               | m- CH <sub>3</sub>        | -NH(CH <sub>2</sub> ) <sub>2</sub> N(C <sub>2</sub> H <sub>5</sub> ) <sub>2</sub>    | 0.9339              |
| 64* | -               | p- CH <sub>3</sub>        | -NH(CH <sub>2</sub> ) <sub>2</sub> N(C <sub>2</sub> H <sub>5</sub> ) <sub>2</sub>    | 0.9057              |
| 65  | -               | m-OCH <sub>3</sub>        | -NH(CH <sub>2</sub> ) <sub>2</sub> N(C <sub>2</sub> H <sub>5</sub> ) <sub>2</sub>    | 0.9289              |
| 66  | -               | p-OCH <sub>3</sub>        | -NH(CH <sub>2</sub> ) <sub>2</sub> N(C <sub>2</sub> H <sub>5</sub> ) <sub>2</sub>    | 0.9561              |
| 67  | -               | p-F                       | -NH(CH <sub>2</sub> ) <sub>2</sub> N(C <sub>2</sub> H <sub>5</sub> ) <sub>2</sub>    | 0.9854              |
| 68  | -               | m-Cl                      | -NH(CH <sub>2</sub> ) <sub>2</sub> N(C <sub>2</sub> H <sub>5</sub> ) <sub>2</sub>    | 0.94                |
| 69  | -               | p-Cl                      | -NH(CH <sub>2</sub> ) <sub>2</sub> N(C <sub>2</sub> H <sub>5</sub> ) <sub>2</sub>    | 0.9765              |
| 70  | -               | p-Br                      | -NH(CH <sub>2</sub> ) <sub>2</sub> N(C <sub>2</sub> H <sub>5</sub> ) <sub>2</sub>    | 0.9885              |
| 71* | -               | 2,3- di(CH <sub>3</sub> ) | -NH(CH <sub>2</sub> ) <sub>2</sub> N(C <sub>2</sub> H <sub>5</sub> ) <sub>2</sub>    | 0.9405              |
| 72  | -               | 2,4- di(CH <sub>3</sub> ) | -NH(CH <sub>2</sub> ) <sub>2</sub> N(C <sub>2</sub> H <sub>5</sub> ) <sub>2</sub>    | 0.9415              |
| 73  | -               | H                         | -NH(CH <sub>2</sub> ) <sub>3</sub> N(C <sub>2</sub> H <sub>5</sub> ) <sub>2</sub>    | 0.8859              |
| 74  | CH <sub>2</sub> | H                         | -NH(CH <sub>2</sub> ) <sub>3</sub> N(C <sub>2</sub> H <sub>5</sub> ) <sub>2</sub>    | 0.9164              |
| 75  | -               | m- CH <sub>3</sub>        | -NH(CH <sub>2</sub> ) <sub>3</sub> N(C <sub>2</sub> H <sub>5</sub> ) <sub>2</sub>    | 0.9779              |
| 76* | -               | p- CH <sub>3</sub>        | -NH(CH <sub>2</sub> ) <sub>3</sub> N(C <sub>2</sub> H <sub>5</sub> ) <sub>2</sub>    | 0.9289              |
| 77  | -               | m-OCH <sub>3</sub>        | -NH(CH <sub>2</sub> ) <sub>3</sub> N(C <sub>2</sub> H <sub>5</sub> ) <sub>2</sub>    | 0.9434              |
| 78  | -               | p-OCH <sub>3</sub>        | -NH(CH <sub>2</sub> ) <sub>3</sub> N(C <sub>2</sub> H <sub>5</sub> ) <sub>2</sub>    | 0.9276              |
| 79* | -               | p-F                       | -NH(CH <sub>2</sub> ) <sub>3</sub> N(C <sub>2</sub> H <sub>5</sub> ) <sub>2</sub>    | 0.9786              |
| 80  | -               | m-Cl                      | -NH(CH <sub>2</sub> ) <sub>3</sub> N(C <sub>2</sub> H <sub>5</sub> ) <sub>2</sub>    | 0.9508              |
| 81* | -               | p-Cl                      | -NH(CH <sub>2</sub> ) <sub>3</sub> N(C <sub>2</sub> H <sub>5</sub> ) <sub>2</sub>    | 0.9774              |
| 82  | -               | p-Br                      | -NH(CH <sub>2</sub> ) <sub>3</sub> N(C <sub>2</sub> H <sub>5</sub> ) <sub>2</sub>    | 0.989               |
| 83  | -               | 2,3- di(CH <sub>3</sub> ) | -NH(CH <sub>2</sub> ) <sub>3</sub> N(C <sub>2</sub> H <sub>5</sub> ) <sub>2</sub>    | 0.9552              |
| 84  | -               | 2,4- di(CH <sub>3</sub> ) | -NH(CH <sub>2</sub> ) <sub>3</sub> N(C <sub>2</sub> H <sub>5</sub> ) <sub>2</sub>    | 0.938               |
| 85  | -               | H                         |  | 0.9175              |
| 86* | CH <sub>2</sub> | H                         |  | 0.9477              |
| 87  | -               | m- CH <sub>3</sub>        |  | 0.9772              |
| 88  | -               | p- CH <sub>3</sub>        |  | 0.9115              |
| 89  | -               | m-OCH <sub>3</sub>        |  | 0.9484              |
| 90* | -               | p-OCH <sub>3</sub>        |  | 0.938               |
| 91* | -               | p-F                       |  | 0.9863              |
| 92  | -               | m-Cl                      |  | 0.9647              |
| 93* | -               | p-Cl                      |  | 0.9175              |

Table 1 Contd...

| No.  | X               | R                         | R <sub>1</sub> | log pA <sub>2</sub> |
|------|-----------------|---------------------------|----------------|---------------------|
| 94   | -               | p-Br                      |                | 0.9885              |
| 95   | -               | 2,3- di(CH <sub>3</sub> ) |                | 0.9385              |
| 96*  | -               | 2,4- di(CH <sub>3</sub> ) |                | 0.9329              |
| 97*  | -               | H                         |                | 0.8764              |
| 98   | CH <sub>2</sub> | H                         |                | 0.9489              |
| 99   | -               | m- CH <sub>3</sub>        |                | 0.9175              |
| 100* | -               | p- CH <sub>3</sub>        |                | 0.9566              |
| 101  | -               | m-OCH <sub>3</sub>        |                | 0.9277              |
| 102  | -               | p-OCH <sub>3</sub>        |                | 0.9656              |
| 103  | -               | p-F                       |                | 0.9604              |
| 104* | -               | m-Cl                      |                | 0.9434              |
| 105  | -               | p-Cl                      |                | 0.9599              |
| 106  | -               | p-Br                      |                | 0.9758              |
| 107  | -               | 2,3- di(CH <sub>3</sub> ) |                | 0.9375              |
| 108  | -               | 2,4- di(CH <sub>3</sub> ) |                | 0.9319              |

\* Test set compounds

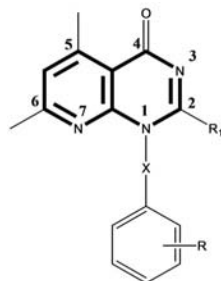
Some of the compounds were found to be more potent than standard drug cetirizine with very low sedative potential and were found to be devoid of anticholinergic side-effects. In order to obtain the most potent and safe molecule in the series, for further lead modification and lead optimization 3D-QSAR study was performed using Comparative molecular field analysis (CoMFA) which may serve to provide useful information on the structural requirements for the antagonist action in terms of contour maps to aid the design of H<sub>1</sub>-receptor antagonists prior to synthesis.

## Results and Discussion

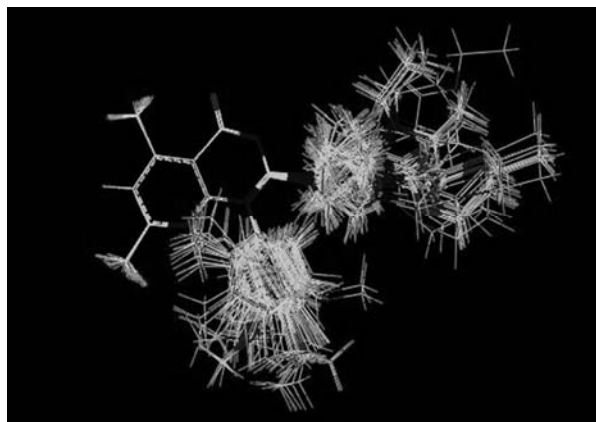
CoMFA technique was used in order to derive stable 3D QSAR model for 108 N<sub>1</sub>-(substituted)aryl-2-(substituted)-5,7- dimethylpyrido [2,3-*d*] pyrimidin- 4 (3*H*)-one H<sub>1</sub> inhibitors. In order to examine the predictive ability and robustness of the CoMFA model, the whole data set of 108 compounds (Table 1) was divided into two parts- the training set consists of 70 compounds and the test set, selected randomly, comprised of 38 compounds. The pA<sub>2</sub> values of the molecules were converted into corresponding

logarithmic values and used as dependent variables. The produced QSAR model was further validated by performing a test set prediction using 38 molecules.

Table 1 lists all structures used in the training and test sets and their logarithmic  $pA_2$  values. Among the used molecules for the QSAR study, the highly bioactive compound 94 was selected as a template molecule. The selected atoms of the template which are common in all studied compounds for the superimposition during the alignment were N1, C2, N3, C4, C5, C6 and N7 as shown in Fig. 1. Fig. 2 illustrates the superimposition of all the molecules.



**Fig. 1** Atom definition of pyridopyrimidine derivatives used for 3D-QSAR study



**Fig. 2** Superimposition of pyridopyrimidine derivatives (training and test set)

The statistics of the final CoMFA model obtained from the PLS analysis, listed in Table 2. The developed CoMFA model gave a significant cross validated regression value of 0.491 with a standard error of prediction of 0.127. This partial least squares analysis suggested that the number of

components to use in the non-cross validated regression model was eight. The resulting non-cross validated linear regression model gave a value of 0.894 and a standard error of estimate of 0.008. The relative contributions of the CoMFA variables to the model were also listed. These values constitute useful information to better understand to what extent the various steric and electrostatic regions can influence the binding of the ligand to H<sub>1</sub> receptor. The fact that the electrostatic contribution is only slightly higher than the steric one (60% vs 40%, respectively) means that the antagonistic activities of compounds for H<sub>1</sub> receptor are highly dependent on the electron density and molecular charges of the ligand.

**Table 2** Statistical results for the best CoMFA model obtained for Pyridopyrimidine derivatives

|                            |        |
|----------------------------|--------|
| $q^2$                      | 0.491  |
| SEP                        | 0.127  |
| n                          | 8      |
| $r^2$                      | 0.894  |
| SEE                        | 0.008  |
| F value (n1=8, n2=61)      | 64.306 |
| Steric contribution        | 0.400  |
| Electrostatic contribution | 0.600  |
| $r^2$ predicted            | 0.624  |

$q^2$  : LOO cross-validated correlation coefficient;  
 SEP : Standard error of prediction;  
 $r^2$  : non-cross-validated correlation coefficient;  
 n : number of components used in the PLS analysis;  
 SEE : standard error estimate;  
 F-value : F-statistic for the analysis.

The predictive ability of the model was determined from the test set of 38 compounds that were not included in the training set. These molecules were aligned in the same way as those in the training set and their activities were predicted by PLS analysis. Table 3 summarizes the predicted results obtained from the CoMFA model for training and test set. The plots of predicted versus actual activity for the training and test set inhibitors are shown in Fig. 3. By comparison of the experimentally observed and theoretically predicted  $\log pA_2$  values of a series of pyridopyrimidine derivatives, it can be seen that CoMFA model performed well in the prediction of the activities of the test inhibitors. In almost all the cases, the predicted values were close to the observed  $\log pA_2$  values, deviating by less than small logarithm unit.

**Table 3** Experimental versus predicted activities ( $\log pA_2$ ), with residual (s), for the pyridopyrimidine derivatives of the training and test set.

| Compound No. | Experimental $\log pA_2$ | Predicted $\log pA_2$ | Residual |
|--------------|--------------------------|-----------------------|----------|
| 1*           | 0.9148                   | 0.905                 | 0.0098   |
| 2            | 0.9375                   | 0.927                 | 0.0105   |
| 3            | 0.9079                   | 0.897                 | 0.0109   |
| 4            | 0.8779                   | 0.907                 | -0.0291  |
| 5*           | 0.9498                   | 0.955                 | -0.0052  |

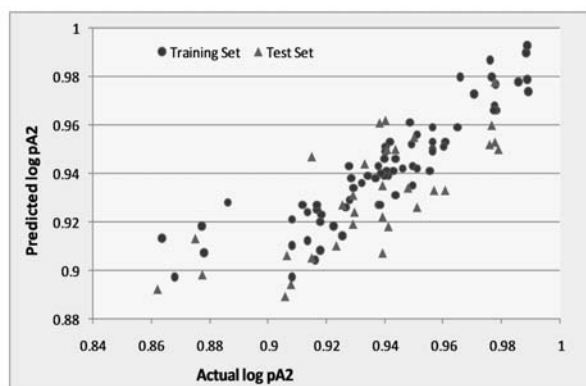
**Table 1** Contd...

| Compound No. | Experimental log pA <sub>2</sub> | Predicted log pA <sub>2</sub> | Residual |
|--------------|----------------------------------|-------------------------------|----------|
| 6            | 0.9164                           | 0.925                         | -0.0086  |
| 7            | 0.9434                           | 0.931                         | 0.0124   |
| 8            | 0.9493                           | 0.943                         | 0.0063   |
| 9            | 0.9253                           | 0.914                         | 0.0113   |
| 10           | 0.9703                           | 0.973                         | -0.0027  |
| 11           | 0.941                            | 0.939                         | 0.002    |
| 12*          | 0.939                            | 0.907                         | 0.032    |
| 13*          | 0.875                            | 0.913                         | -0.038   |
| 14           | 0.8633                           | 0.913                         | -0.0497  |
| 15           | 0.9493                           | 0.935                         | 0.0143   |
| 16*          | 0.941                            | 0.918                         | 0.023    |
| 17           | 0.9508                           | 0.956                         | -0.0052  |
| 18*          | 0.939                            | 0.935                         | 0.004    |
| 19*          | 0.9532                           | 0.918                         | 0.0352   |
| 20           | 0.9405                           | 0.941                         | -0.0005  |
| 21           | 0.94                             | 0.951                         | -0.011   |
| 22*          | 0.9774                           | 0.953                         | 0.0244   |
| 23           | 0.9079                           | 0.921                         | -0.0131  |
| 24*          | 0.8774                           | 0.898                         | -0.0206  |
| 25           | 0.918                            | 0.923                         | -0.005   |
| 26           | 0.9159                           | 0.904                         | 0.0119   |
| 27*          | 0.9232                           | 0.91                          | 0.0132   |
| 28           | 0.9132                           | 0.924                         | -0.0108  |
| 29*          | 0.9756                           | 0.952                         | 0.0236   |
| 30           | 0.9426                           | 0.941                         | 0.0016   |
| 31*          | 0.9604                           | 0.933                         | 0.0274   |
| 32           | 0.9561                           | 0.95                          | 0.0061   |
| 33           | 0.9365                           | 0.938                         | -0.0015  |
| 34           | 0.9777                           | 0.977                         | 0.0007   |
| 35*          | 0.9164                           | 0.949                         | -0.0326  |
| 36*          | 0.9294                           | 0.924                         | 0.0054   |
| 37           | 0.8677                           | 0.897                         | -0.0293  |
| 38*          | 0.9063                           | 0.906                         | 0.0003   |
| 39           | 0.9077                           | 0.894                         | 0.0137   |
| 40*          | 0.9776                           | 0.901                         | 0.0766   |
| 41           | 0.9263                           | 0.926                         | 0.0003   |
| 42           | 0.9253                           | 0.927                         | -0.0017  |
| 43*          | 0.9518                           | 0.908                         | 0.0438   |
| 44*          | 0.8621                           | 0.892                         | -0.0299  |
| 45*          | 0.9222                           | 0.918                         | 0.0042   |
| 46           | 0.9561                           | 0.953                         | 0.0031   |
| 47           | 0.9395                           | 0.939                         | 0.0005   |
| 48           | 0.9289                           | 0.919                         | 0.0099   |
| 49*          | 0.877                            | 0.918                         | -0.041   |
| 50           | 0.9079                           | 0.91                          | -0.0021  |
| 51           | 0.9132                           | 0.912                         | 0.0012   |
| 52*          | 0.939                            | 0.922                         | 0.017    |
| 53           | 0.9459                           | 0.942                         | 0.0039   |
| 54           | 0.9283                           | 0.938                         | -0.0097  |
| 55*          | 0.9508                           | 0.926                         | 0.0248   |
| 56           | 0.9561                           | 0.949                         | 0.0071   |
| 57*          | 0.9148                           | 0.947                         | -0.0322  |
| 58           | 0.9774                           | 0.968                         | 0.0094   |
| 59*          | 0.94                             | 0.962                         | -0.022   |
| 60           | 0.9395                           | 0.946                         | -0.0065  |
| 61*          | 0.9763                           | 0.96                          | 0.0163   |
| 62           | 0.9881                           | 0.99                          | -0.0019  |
| 63           | 0.9339                           | 0.939                         | -0.0051  |
| 64*          | 0.9057                           | 0.889                         | 0.0167   |
| 65           | 0.9289                           | 0.934                         | -0.0051  |
| 66           | 0.9561                           | 0.959                         | -0.0029  |
| 67           | 0.9854                           | 0.978                         | 0.0074   |

Table 1 Contd...

| Compound No. | Experimental log pA <sub>2</sub> | Predicted log pA <sub>2</sub> | Residual |
|--------------|----------------------------------|-------------------------------|----------|
| 68           | 0.94                             | 0.949                         | -0.009   |
| 69           | 0.9765                           | 0.98                          | -0.0035  |
| 70           | 0.9885                           | 0.993                         | -0.0045  |
| 71*          | 0.9405                           | 0.95                          | -0.0095  |
| 72           | 0.9415                           | 0.953                         | -0.0115  |
| 73           | 0.8859                           | 0.928                         | -0.0421  |
| 74           | 0.9164                           | 0.927                         | -0.0106  |
| 75           | 0.9779                           | 0.966                         | 0.0119   |
| 76*          | 0.9289                           | 0.931                         | -0.0021  |
| 77           | 0.9434                           | 0.946                         | -0.0026  |
| 78           | 0.9276                           | 0.943                         | -0.0154  |
| 79*          | 0.9786                           | 0.95                          | 0.0286   |
| 80           | 0.9508                           | 0.942                         | 0.0088   |
| 81*          | 0.9774                           | 0.978                         | -0.0006  |
| 82           | 0.989                            | 0.974                         | 0.015    |
| 83           | 0.9552                           | 0.941                         | 0.0142   |
| 84           | 0.938                            | 0.927                         | 0.011    |
| 85           | 0.9175                           | 0.92                          | -0.0025  |
| 86*          | 0.9477                           | 0.934                         | 0.0137   |
| 87           | 0.9772                           | 0.966                         | 0.0112   |
| 88           | 0.9115                           | 0.927                         | -0.0155  |
| 89           | 0.9484                           | 0.961                         | -0.0126  |
| 90*          | 0.938                            | 0.961                         | -0.023   |
| 91*          | 0.9863                           | 0.941                         | 0.0453   |
| 92           | 0.9647                           | 0.959                         | 0.0057   |
| 93*          | 0.9175                           | 0.97                          | -0.0525  |
| 94           | 0.9885                           | 0.979                         | 0.0095   |
| 95           | 0.9385                           | 0.94                          | -0.0015  |
| 96*          | 0.9329                           | 0.944                         | -0.0111  |
| 97*          | 0.8764                           | 0.929                         | -0.0526  |
| 98           | 0.9489                           | 0.952                         | -0.0031  |
| 99           | 0.9175                           | 0.908                         | 0.0095   |
| 100*         | 0.9566                           | 0.933                         | 0.0236   |
| 101          | 0.9277                           | 0.929                         | -0.0013  |
| 102          | 0.9656                           | 0.98                          | -0.0144  |
| 103          | 0.9604                           | 0.953                         | 0.0074   |
| 104*         | 0.9434                           | 0.95                          | -0.0066  |
| 105          | 0.9599                           | 0.951                         | 0.0089   |
| 106          | 0.9758                           | 0.987                         | -0.0112  |
| 107          | 0.9375                           | 0.943                         | -0.0055  |
| 108          | 0.9319                           | 0.936                         | -0.0041  |

\* Test set compounds



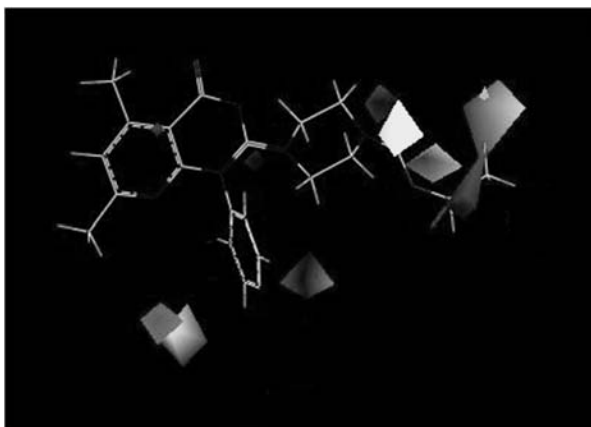
**Fig. 3** CoMFA predicted versus actual log pA<sub>2</sub> values for the training and test set. Correlation coefficient ( $r^2$ ) = 0.894 for training set and 0.624 for test set.

The predictive correlation coefficient ( $r^2_{pred}$ ) is defined as  $r^2_{pred} = (SD - PRESS) / SD$  where SD is the sum of squared deviations between the biological activities of the test set and mean activity of the training set molecules and PRESS is the sum of squared deviation between predicted and actual activity values for every molecule in the test set. The  $r^2_{pred}$  obtained was 0.624 for the CoMFA model (Table 2) which indicates good external predictive power of the model. All results including statistical data from this predictive validation analysis further confirmed the robustness of the obtained model in terms of their predictive ability.

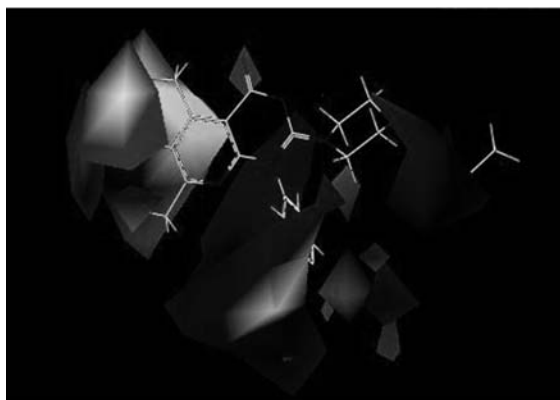
From the validated CoMFA model, it is possible to identify regions of space around the molecules where changes with altering substitution would lead to an increase or a decrease biological activity. This was achieved by using a CoMFA



contour map for the model, as shown in Fig. 4 and 5. Fig. 4 shows the steric contour map for the CoMFA model with one of the highly active inhibitor 94 ( $\log pA_2 = 0.9885$ ) as a reference. The steric field defined by the green colored contours represents regions of high steric tolerance, while yellow colored contours represent regions of unfavorable steric effect. Fig. 5 shows the electrostatic contour map where the blue colored contours represent the regions where positively charged groups enhance the activity and red colored contours where the negatively charged groups enhance the activity.



**Fig. 4** Standard coefficient steric contour map of final CoMFA analysis. Green contours (80% contribution) refer to sterically favored regions and yellow counters (20% contribution) are sterically disfavored.



**Fig. 5** Standard coefficient electrostatic contour map of final CoMFA analysis. Blue contours (80% contribution) refer to regions where positively charged substituents are favored; red contours (20% contribution) indicate regions where negatively charged substituents are favored.

The CoMFA steric map displays green colored contours bordering the terminal end of  $R_1$  substituent while yellow colored contours are mapped on the carbonyl group of  $R_1$  (bulky group disfavored) in the reference molecule.

CoMFA electrostatic map displays blue colored contours around the carbonyl group of  $R_1$  (negative charged groups disfavored) explaining the decreased activity. On the other side, green contours around  $R_1$  indicate that length and bulk of tail of the side chain play a critical role in determining the biological activity as long chain is favorable for the activity with electropositive atoms. In the case of highly active compounds **70** ( $R_1 = -NH(CH_2)_2N(C_2H_5)_2$ ,  $\log pA_2 = 0.9885$ ) and **82** ( $R_1 = -NH(CH_2)_3N(C_2H_5)_2$ ,  $\log pA_2 = 0.989$ ), long and bulky substituent as  $R_1$  without any carbonyl function resulted in good activity in comparison to other molecules such as compound **14** ( $R_1 =$  piperazine ring,  $\log pA_2 = 0.8633$ ) and **37,44** ( $R_1 =$  N,N-dimethyl,  $\log pA_2 = 0.8677$  and  $0.8621$  respectively).

Green colored contours at the *para* position of the aryl ring indicate that bulky substitutions favor the activity. In every series, bulky halogens at *para* position (Compound **10**, **22** and **70**) gave higher activity compared to smaller  $-CH_3$  or  $-H$  groups (Compound **4**, **14** and **37**).

The CoMFA electrostatic contour map displays big zones around the whole 5,7-dimethylpyrido[2,3-*d*]pyrimidin ring. The blue contour around the aryl ring indicates that electropositive groups at any positions of phenyl ring increase the activity. The *para* position where a small red contour supports the conclusion that electronegative halogens increase the activity.

Red contour around the 5-methyl group of the 5,7-dimethylpyrido[2,3-*d*]pyrimidin ring suggests that instead of methyl more electronegative groups like halogens, oxygen, sulfur etc. can result in better activity profile.

Based on the results from CoMFA maps, it is possible to assume that the common pyridopyrimidine motif present in the derivatives is the most important part for the activity of this series. Furthermore, the length and bulk of  $R_1$  side chain as well as electronic environment of aryl ring are the other influencing factors. The information obtained from the study provides a means for predicting the activity of this series of compounds for guiding further structural modifications and synthesis of potent  $H_1$  antagonists.

## Materials and Methods

### Data Set and Molecular Modeling

The QSAR modeling analyses, calculations, and visualizations for CoMFA were performed using the SYBYL X1.2 package<sup>9</sup> running on windows based workstation. To obtain reliable results, the total 108  $H_1$  inhibitors of a  $N_1$ -(substituted)aryl-2-(substituted)-5,7-dimethylpyrido[2,3-*d*]pyrimidin-4(3*H*)-one series<sup>8</sup> were divided into two sets, the training set consisting of 70 compounds and the test set consisting of 38 compounds (Table 1). Selection of training set and test set molecules was done by considering the fact that thirty eight compounds of the test set cover approximately all range of

log p<sub>A2</sub> similar to that of training set. Thus, the test set was truly representative of the training set. The 3D structures of the compounds were created in MOL2 format using the sketcher module of the SYBYL program. Gasteiger-Huckel charges were calculated for all compounds and then energy minimization was done using the Tripos force field.

### Alignment of Molecules

One of the most important parameters in CoMFA is the relative alignment of all the compounds to one another so that they have a comparable conformation and a similar orientation of pharmacophoric groups in space<sup>10</sup>. In this study, the most active compound 94 was used as a template for superimposition, assuming that its lowest-energy conformation represents the most probable bioactive conformation of the pyridopyrimidine analogs at the H<sub>1</sub> receptor. The minimized molecules were superimposed (Fig. 2) by the atom-fit method choosing the seven features numbered from 1 to 7 in 5,7-dimethylpyrido(2,3-*d*)pyrimidin-4(3*H*)-one fragment (Fig.1).

### PLS Analysis

CoMFA is a popular 3D-QSAR technique which uses Partial least square (PLS) as the data analysis method. PLS is then used to derive an equation that correlates the biological activity with the different probes at various lattice points<sup>11</sup>. Default values provided in the Tripos CoMFA module were used with 2.0 Å grid spacing using a sp<sup>3</sup> carbon atom with a +1 point charge as a probe to explore the steric and electrostatic interactions at the lattice points in the grid. The default cut-off value was set at 30 kcal/mol. Leave-one-out cross-validated PLS analysis was performed to determine the optimal number of components by a minimum predictive sum of squares (PRESS) value and maximum q<sup>2</sup>. The q<sup>2</sup> statistic is defined as  $q^2 = 1 - \frac{\sum (Y_{\text{pred}} - Y_{\text{actual}})^2}{\sum (Y_{\text{actual}} - Y_{\text{mean}})^2}$  where Y<sub>pred</sub>, Y<sub>actual</sub>, and Y<sub>mean</sub> are the predicted, actual and mean values of the target property respectively.  $\sum (Y_{\text{pred}} - Y_{\text{actual}})^2$  is the PRESS. Cross-validation of the dependent column and the CoMFA column was performed with 2.0 kcal/mol column filtering to minimize the influence of noisy columns and scaled by the CoMFA standard deviation. The q<sup>2</sup> values obtained were considered significant at 0.3. The optimal number of components was used to generate the final Non-cross validation PLS model and the 3D-QSAR model was used to predict the target properties of the compounds in the testing set. Non-cross validated (r<sup>2</sup>) values were determined for the models using linear regression analysis (with variances reported as the standard error of estimation, SEE.) which were considered significant when r<sup>2</sup> is greater than 0.7. The 3D graphical representation of the steric and

electrostatic fields generated through CoMFA are shown with the relative contributions represented as a 3D coefficient map with favored 80% steric (green) and electrostatic (blue) effects and 20% disfavored steric (yellow) and electrostatic effects (red). Green colored areas of the map indicate where sterically bulky groups may enhance interaction affinity. Blue colored areas (80%) indicate regions where a more positively charged group will likely lead to increased binding affinity, while red areas indicate where a more negatively charged group will likely lead to increased binding (20%).

### Conclusion

In present study, the 3D-QSAR model, CoMFA was applied to predict the Histamine H<sub>1</sub> receptor antagonist activity of a series of substituted pyridopyrimidine derivatives. The best model had satisfactory statistical results in terms of q<sup>2</sup> and r<sup>2</sup> values, and showed excellent predictivity of the test set, in the external validation, showing no outliers. Information obtained from CoMFA maps gave good insight about the influence of various structural features for the H<sub>1</sub> antagonist activity and may be considered as a powerful tool in designing and forecasting more efficacious analogs.

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