ENERGY & EXERGY ANALYSIS OF THERMAL POWER PLANT

AT WANAKBORI THERMAL POWER STATION

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- Energy and Exergy analysis

Introduction

- Energy sector plays very important role developing country's economic development.
- Energy consumption shows the development stages of country and living standards of communities.
- India is third largest energy consumer in world as per energy information administration of India.
- Objective of energy conservation is to reduce energy generation cost.
- Electricity generation capacity 1,362 MW in 1947 to 3,56,100 MW in year 2020.
- Thermal power plant generation capacity 854 MW in 1947 to 2,26,279 in year 2020.

Electricity Installed Capacity of India

- Coal 2,05,134.5 MW
- Hydro 50,382.38 MW
- Wind 37,693.75 MW
- Solar 34,627.82 MW
- Biomass 10,022.95 MW

Coal

- Nuclear 6,780 MW
- Gas 24,955.36 MW
- Diesel 509.71 MW



Motivation

- The rate of power demand increases is extremely high compared with the rate of generation capacity increase.
- To establish new power plant required few years. However, improving performance of present power plant certain demand can be fulfilled with less time period.
- To improve capacity of the plant one needs to determine performance of various components.
- Fore that purpose component wise energy and exergy analysis will help to identify the performance of the power plant.

Energy

- Energy is define as the capacity of a physical system to perform work. Classification of energy:
- 1. Primary and Secondary energy
- 2. Commercial and Non-commercial energy
- 3. Renewable and Non-Renewable energy

Exergy

- Exergy is define as the maximum useful work possible during a process.
- After system and surroundings reach equilibrium, the exergy is zero.

Classification of exergy : Physical and Chemical



Coal use in Wanakbori Thermal Power Plant

- In Wanakbori Thermal Power Plant we use mainly 3 types of coal is used.
 - Indian coal
 - Wash coal
 - Imported coal
- Indian coal: Bituminous type Indian coal is use at GSECL Wanakbori thermal power plan. Its calorific value is nearby 3200 kCal/kg. some of Indian coal is very good equality Thames calorific value is also around 4700 kCal/kg.
- Wash coal: Coal comes from mine is a complex of materials. By washing quality of coal is improved. Impurities as Sulfur, Ash and Rock are removed from coal so transport charges are reduced. Its calorific value is around 4000 kCal/kg.
- Imported coal: It is very good quality of coal. Its calorific value is around 5500 kCal/kg and sometime more than 6000 kCal/kg.

Analysis of coal

• The composition of the coal varies from mine to mine. Here it is necessary to analysis for commercial purpose and price fixing.



Proximate analysis and its methodology

Content	With/without Lid	Apparatus	Temperature	Time
Percentage of moisture (M)	Without Lid	Oven	110 °C	One Hour
Percentage of volatile matter (VM)	With Lid	Muffle Furnace	925 °C	Seven Minutes
Percentage of ash (A)	Without lid	Muffle Furnace	700 °C	Half an Hour
Percentage of carbon (C)	C=100 – M – VM - A			

Proximate Analysis Results



Calculation of Boiler efficiency

- Thermal efficiency of boiler is defined as the percentage of heat input that is effectively utilized to generation steam. There are two methods of assessing boiler efficiency.
- It is not always true that a boiler will work at its rated efficiency. Almost all the times, it has been found that the boilers operate at much lower than the rated efficiencies if proper efficiency monitoring is not done.



Direct Method

• This is also known an 'input output method' due to the fact that it need only the useful output (steam) and the heat input (Fuel) for evaluating the efficiency. This efficiency can be evaluated using the formula.

Boiler Efficiency= $\frac{\text{Steam Generation } *(\text{ Steam Enthalpy} - \text{Feed Water Enthalpy})}{\text{FuelConsumption } * \text{ CalorificValue}}$

- Quickly find the efficiency of boilers
- Few parameters required
- Few instrument required for monitoring
- Does not give clues to the operator as to why efficiency of system is lower
- Does not calculate various losses accountable for various efficiency levels

Input data for Direct Method

Steam generation	656	Ton/h
Enthalpy of super-heated steam (P= T=	3420.32	kJ/kg
Enthalpy of feed water (T=)	1014.72	kJ/kg
Gross Calorific Value of Fuel	16430.05	kJ/kg
Coal consumption	115	Ton/h

Efficiency = 83.52 %

Indirect method

- Indirect method is also called as heat loss method
- The efficiency can be arrived by subtracting the heat loss fractions from 100

The principal loss occur in a boiler are

- Loss of heat due to dry flue gas
- Loss of heat due to Combustible loss
- Loss of heat due to Sensible heat loss
- Loss of heat due to radiation
- Loss of heat due to moisture
- Loss of heat due to mill rejection
- Loss of heat due to carbon monoxide
- \succ Loss of heat due to moisture in air
- Loss of heat due to unburnt

Result of Indirect method

Sr No	Results	Value
1	Dry gas loss	5.708 %
2	Combustible loss	0.634 %
3	Sensible heat loss	0.426 %
4	Radiation loss	0.214 %
5	Moisture loss	6.501 %
6	Loss due to mill rejection	0.019 %
7	Loss due to carbon monoxide (CO)	0.151 %
8	Loss due to moisture in air	0.101 %
9	Unaccounted loss	1 %
10	Total boiler losses	14.754 %
11	Boiler efficiency	85.246 %

Energy Analysis

• Mass balance equation for control volume under steady flow

$$\dot{m}_{in} = \dot{m}_{out}$$

• Energy balance equation for control volume under steady flow

$$\dot{m}_{in}\left(h+\frac{v^2}{2}+gz\right)_{in}+Q=\dot{m}_{out}\left(h+\frac{v^2}{2}+gz\right)_{out}+W$$

• Assumption : Change in K.E. and P.E. is neglected

• First low efficiency or Energy efficiency

$$Energy Efficiency = \frac{Energy Out}{Energy Input} = 1 - \frac{Energy Loss}{Energy Input}$$

Exergy Analysis

Exergy balance equation for control volume under steady flow
 Exergy In = Exergy Utilized + Exergy Loss + Exergy Distruction

$$(ma)_{in} = (ma)_{out} + Q_L \left(1 - \frac{T_0}{T_S}\right) + I$$
$$a = h + \frac{v^2}{2} + gz - T_0S$$

Assumption : Exergy due change in KE and PE is neglected.

• Second low efficiency or Energy efficiency

 $Exergy \ Efficiency = \frac{Exergy \ Out}{Exergy \ Input} = 1 - \frac{Exergy \ Distruction + Exergy \ Loss}{Energy \ Input}$



Economizer Analysis Result

Energy Analysis Result

Exergy Analysis Result



Air Pre-Heater

Figure shows that primary and secondary air enters from section 1 and leaves from section 2. Flue gas enters from 3 and leaves from section 4. In between primary and secondary air gets heat from flue gas and flue gas loss heat energy.



Air Pre-Heater Required Data

Mass flow rate of Primary Air	m1, m2	28.22 kg/s
Temperature of primary air in	T1	308 K
Temperature of primary air out	T2	609 K
Mass flow rate of secondary air	m3, m4	157.94 kg/s
Temperature of secondary air in	Т3	308 K
Temperature of secondary air out	T4	604 K
Mass flow rate of flue gas at Air Pre-heater	m5, m6	258.94 kg/s
Temperature of flue gas in at Air Pre-heater	Т5	630 K
Temperature of flue gas out at Air Pre-heater	Т6	423 K

Energy Analysis of Air Pre-Heatere



Exergy Analysis of Air-Pre Heater



Air Pre-heater Analysis Result

Energy Analysis Result

Exergy Analysis Result



Super Heater

Figure shows that steam enters from section 1 and leaves from section 2. Flue gas enters from section 3 and leaves from section 4. In between steam get sensible heat from flue gas become dry and flue gas loss heat energy.



Super Heater Required Data

Mass flow rate of super heater steam	m1, m2	182.22 kg/s
Temperature of super heater steam in	T1	622 K
Enthalpy of super heater steam in	h1	1658.68 kJ/kg
Entropy of super heater steam in	S1	3.76 kJ/kg K
Temperature of super heater steam out	T2	813 K
Enthalpy of super heater steam out	h2	3417.72 kJ/kg
Entropy of super heater steam out	S2	6.47 kJ/kg K
Mass flow rate of flue gas at super heater	m3, m4	258.94 kg/s
Temperature of flue gas in at super heater	Т3	1436 K
Temperature of flue gas out at super heater	T4	964 K

Energy Analysis of Super Heater



Exergy Analysis of Super Heater



Super Heater Analysis Result

Energy Analysis Result

Exergy Analysis Result



Re-heater

Figure shows that steam enters from section 1 and leaves from section 2. Flue gas enters from section 3 and leaves from section 4. In between steam get sensible heat from flue gas and flue gas loss heat energy.



Re-heater Heater Required Data

Mass flow rate of re-heater steam	m1, m2	157.17 kg/s
Temperature of re-heater steam in	T1	609.6 K
Enthalpy of re-heater steam in	h1	3065.30 kJ/kg
Entropy of re-heater steam in	S1	6.57 kJ/kg K
Temperature of re-heater steam out	T2	813 K
Enthalpy of re-heater steam out	h2	3540.76 kJ/kg
Entropy of re-heater steam out	S2	7.26 kJ/kg K
Mass flow rate of flue gas at re-heater	m3, m4	258.94 kg/s
Temperature of flue gas in at re-heater	Т3	1286 K
Temperature of flue gas out at re-heater	T4	1022 K

Energy Analysis of Re-heater



Exergy Analysis of Super Heater



Re-heater Analysis Result

Energy Analysis Result

Exergy Analysis Result



LPH 1

Figure shows that feed water enters from section 1 and leaves from section 2. Steam enters from section 3 and leaves from section 4. In between water get sensible heat from steam. **3**


LPH 1 Required Data Mass flow rate of LPH 1 feed water in Temperature of LPH 1 feed water in Enthalpy of LPH 1 feed water in Entropy of LPH 1 feed water in Mass flow rate of LPH 1 feed water out Temperature of LPH 1 feed water out Enthalpy of LPH 1 feed water out Entropy of LPH 1 feed water out Mass flow rate of LPH 1 steam in Temperature of LPH 1 steam in Enthalpy of LPH 1 steam in Entropy of LPH 1 steam in Mass flow rate of LPH 1 water out Temperature of LPH 1 water out Enthalpy of LPH 1 water out Entropy of LPH 1 water out

125.25 kg/s 322.4 K 208.46 kJ/kg 0.69 kJ/kg K 125.25 kg/s 346.2 K 307.67 kJ/kg 0.99 kJ/kg K 5.36 kg/s 357.6 K 2652.25 kJ/kg 7.66 kJ/kg K 5.36 kg/s 319.9 K 327.35 kJ/kg 1.04 kJ/kg K

m1

T1

h1

S1

m2

T2

h2

S2

m3

T3

h3

S3

m4

T4

h4

S4

Energy Analysis of LPH 1



Exergy Analysis of LPH 1



LPH 1 Analysis Result

Energy Analysis Result

Exergy Analysis Result



LPH 2

Figure shows that feed water enters from section 1. Steam enters from section 3. hot water enters from section 4. Section 3 and 4 steam and water mix it gives heat to feed water then both become water it will pump in feed water which comes out from section 2. In between feed water get sensible heat from steam and hot water.



LPH 2 Required Data

m1

T1

h1

S1

m2

T2

h2

S2

m3

T3

h3

S3

m4

T4

h4

S4

Mass flow rate of LPH 2 feed water in Temperature of LPH 2 feed water in Enthalpy of LPH 2 feed water in Entropy of LPH 2 feed water in Mass flow rate of LPH 2 feed water out Temperature of LPH 2 feed water out Enthalpy of LPH 2 feed water out Entropy of LPH 2 feed water out Mass flow rate of LPH 2 steam in Temperature of LPH 2 steam in Enthalpy of LPH 2 steam in Entropy of LPH 2 steam in Mass flow rate of LPH 2 hot water in Temperature of LPH 2 hot water in Enthalpy of LPH 2 hot water in Entropy of LPH 2 hot water in

125.25 kg/s 346.2 K 307.67 kJ/kg 0.99 kJ/kg K 143.44 kg/s 373.8 K 423.62 kJ/kg 1.31 kJ/kg K 5.03 kg/s 515.8 K 2959.08 kJ/kg 7.91 kJ/kg K 13.17 kg/s 378.2 K 558.83 kJ/kg <u>1.66 kI/kø K</u>

Energy Analysis of LPH 2





Exergy Analysis of LPH 2

LPH 2 Analysis Result

Energy Analysis Result

Exergy Analysis Result



LPH 3

Figure shows that feed water enters from section 1 and leaves from section 2. Steam enters from section 3. hot water enters from section 4. Section 3 and 4 steam and water mix it gives heat to feed water then both become water. It leaves from section 5. In between feed water get sensible heat from steam and hot water.



LPH 3 Required Data

Mass flow rate of LPH 3 feed water in Temperature of LPH 3 feed water in Enthalpy of LPH 3 feed water in Entropy of LPH 3 feed water in Mass flow rate of LPH 3 feed water out Temperature of LPH 3 feed water out Enthalpy of LPH 3 feed water out Entropy of LPH 3 feed water out Mass flow rate of LPH 3 steam in Temperature of LPH 3 steam in Enthalpy of LPH 3 steam in Entropy of LPH 3 steam in

143.44 kg/s
373.8 K
423.62 kJ/kg
1.31 kJ/kg K
143.44 kg/s
400.9 K
538.74 kJ/kg
1.61 kJ/kg K
6.47 kg/s
531.1 K
2992.99 kJ/kg
7.55 kJ/kg

m1

T1

h1

S1

m2

T2

h2

S2

m3

T3

h3

S3

LPH 3 Required Data

Mass flow rate of LPH 3 hot water in Temperature of LPH 3 hot water in Enthalpy of LPH 3 hot water in Entropy of LPH 3 hot water in Mass flow rate of LPH 3 water out Temperature of LPH 3 water out Enthalpy of LPH 3 water out

6.69 kg/s 405.9 K 675.62 kJ/kg 1.93 kJ/kg K 13.17 kg/s 378.2 K 558.83 kJ/kg 1.66 kJ/kg K

m4

T4

h4

S4

m5

T5

h5

S5

Energy Analysis of LPH 3





Exergy Analysis of LPH 3

LPH 3 Analysis Result

Energy Analysis Result

Exergy Analysis Result



LPH 4

Figure shows that feed water enters from section 1 and leaves from section 2. Steam enters from section 3. Section 3 steam gives heat to feed water then become water. It leaves from section 4. In between feed water get sensible heat from steam.



LPH 4 Required Data

m1

T1

h1

S1

m2

T2

h2

S2

m3

T3

h3

S3

m4

T4

h4

S4

Mass flow rate of LPH 4 feed water in Temperature of LPH 4 feed water in Enthalpy of LPH 4 feed water in Entropy of LPH 4 feed water in Mass flow rate of LPH 4 feed water out Temperature of LPH 4 feed water out Enthalpy of LPH 4 feed water out Entropy of LPH 4 feed water out Mass flow rate of LPH 4 steam in Temperature of LPH 4 steam in Enthalpy of LPH 4 steam in Entropy of LPH 4 steam in Mass flow rate of LPH 4 water out Temperature of LPH 4 water out Enthalpy of LPH 4 water out Entropy of LPH 4 water out

143.44 kg/s 400.9 K 538.74 kJ/kg 1.61 kJ/kg K 143.44 kg/s 428 K 654.69 kJ/kg 1.89 kJ/kg K 6.69 kg/s 622.9 K 3165.03 kJ/kg 7.52 kJ/kg K 6.69 kg/s 405.9 K 675.62 kJ/kg <u>1.93 kI/kø K</u>

Energy Analysis of LPH 4



Exergy Analysis of LPH 4



LPH 4 Analysis Result

Energy Analysis Result

Exergy Analysis Result



HPH 1

Figure shows that feed water enters from section 1 and leaves from section 2. Steam enters from section 3. hot water enters from section 4. Section 3 and 4 steam and water mix it gives heat to feed water then both become water. It leaves from section 5. In between feed water get sensible heat from steam and hot water.



HPH 1 Required Data

Mass flow rate of HPH 1 feed water in Temperature of HPH 1 feed water in Enthalpy of HPH 1 feed water in Entropy of HPH 1 feed water in Mass flow rate of HPH 1 feed water out Temperature of HPH 1 feed water out Enthalpy of HPH 1 feed water out Entropy of HPH 1 feed water out Mass flow rate of HPH 1 steam in Temperature of HPH 1 steam in Enthalpy of HPH 1 steam in Entropy of HPH 1 steam in

1	69.72 kg/s
Z	43.4 K
-	730.04 kJ/kg
	2.02 kJ/kg K
1	69.72 kg/s
Z	58.4 K
-	795.34 kJ/kg
	2.17 kJ/kg K
Ę	5.06 kg/s
-	710.1 K
	3340.43 kJ/kg
-	7.49 kJ/kg

m1

T1

h1

S1

m2

T2

h2

S2

m3

T3

h3

S3

HPH 1 Required Data

Mass flow rate of HPH 1 hot water in Temperature of HPH 1 hot water in Enthalpy of HPH 1 hot water in Entropy of HPH 1 hot water in Mass flow rate of HPH 1 water out Temperature of HPH 1 water out Enthalpy of HPH 1 water out 21.17 kg/s 460.4 K 832.18 kJ/kg 2.30 kJ/kg K 25.19 kg/s 437.2 K 795.34 kJ/kg 2.22 kJ/kg K

m4

T4

h4

S4

m5

T5

h5

S5

Energy Analysis of HPH 1



Exergy Analysis of HPH 1



HPH 1 Analysis Result

Energy Analysis Result

Exergy Analysis Result



HPH 2

Figure shows that feed water enters from section 1 and leaves from section 2. Steam enters from section 3. hot water enters from section 4. Section 3 and 4 steam and water mix it gives heat to feed water then both become water. It leaves from section 5. In between feed water get sensible heat from steam and hot water.



HPH 2 Required Data

Mass flow rate of HPH 2 feed water in Temperature of HPH 2 feed water in Enthalpy of HPH 2 feed water in Entropy of HPH 2 feed water in Mass flow rate of HPH 2 feed water out Temperature of HPH 2 feed water out Enthalpy of HPH 2 feed water out Entropy of HPH 2 feed water out Mass flow rate of HPH 2 steam in Temperature of HPH 2 steam in Enthalpy of HPH 2 steam in Entropy of HPH 2 steam in

169.72 kg/s
458.4 K
795.34 kJ/kg
2.17 kJ/kg K
169.72 kg/s
495.2 K
958.18 kJ/kg
2.51 kJ/kg K
11.86 kg/s
582.2 K
3.34.85 kJ/kg
6.65 kJ/kg

m1

T1

h1

S1

m2

T2

h2

S2

m3

T3

h3

S3

HPH 2 Required Data

Mass flow rate of HPH 2 hot water in Temperature of HPH 2 hot water in Enthalpy of HPH 2 hot water in Entropy of HPH 2 hot water in Mass flow rate of HPH 2 water out Temperature of HPH 2 water out Enthalpy of HPH 2 water out

9.28 kg/s 499.7 K 1000.45 kJ/kg 2.62 kJ/kg K 21.14 kg/s 460.4 K 832.18 kJ/kg 2.30 kJ/kg K

m4

T4

h4

S4

m5

T5

h5

S5

Energy Analysis of HPH 2



Exergy Analysis of HPH 2



HPH 2 Analysis Result

Energy Analysis Result

Exergy Analysis Result



HPH 3

Figure shows that feed water enters from section 1 and leaves from section 2. Steam enters from section 3. Section 3 steam gives heat to feed water then become water. It leaves from section 4. In between feed water get sensible heat from steam.



HPH 3 Required Data

m1

T1

h1

S1

m2

T2

h2

S2

m3

T3

h3

S3

m4

T4

h4

<u>S4</u>

Mass flow rate of HPH 3 feed water in Temperature of HPH 3 feed water in Enthalpy of HPH 3 feed water in Entropy of HPH 3 feed water in Mass flow rate of HPH 3 feed water out Temperature of HPH 3 feed water out Enthalpy of HPH 3 feed water out Entropy of HPH 3 feed water out Mass flow rate of HPH 3 steam in Temperature of HPH 3 steam in Enthalpy of HPH 3 steam in Entropy of HPH 3 steam in Mass flow rate of HPH 3 water out Temperature of HPH 3 water out Enthalpy of HPH 3 water out Entropy of HPH 3 water out

169.72 kg/s 495.2 K 958.18 kJ/kg 2.51 kJ/kg K 169.72 kg/s 520.4 K 1074.13 kJ/kg 2.74 kJ/kg K 9.28 kg/s 633.4 K 3122.76 kJ/kg 6.62 kJ/kg K 9.28 kg/s 499.7 K 1000.45 kJ/kg 2.62 kJ/kg K

Energy Analysis of HPH 3



Exergy Analysis of HPH 3


HPH 3 Analysis Result

Energy Analysis Result



Turbine

Figure shows that super-heated steam in HP turbine enters from section 1, reheated steam in IP turbine enters from section 2, Steam in LP turbine enters from section 3. Steam leaves by extraction in HP turbine from section 4 and section 5. Steam leaves in HP turbine from section 6. Steam leaves by extraction in IP turbine from section 7, section 8, section 9 and section 10. Steam leaves in IP turbine from section 11. Steam leaves by extraction in LP turbine by section 12. Steam leaves in IP turbine from section 13. In between thermal energy are converted to mechanical work.



m1

T1

h1

S1

m2

T2

h2

S2

m3

T3

h3

S3

Mass flow rate of HP turbine steam in Temperature of HP turbine steam in Enthalpy of HP turbine steam in Entropy of HP turbine steam in Mass flow rate of IP turbine steam in Temperature of IP turbine steam in Enthalpy of IP turbine steam in Entropy of IP turbine steam in Mass flow rate of LP turbine steam in Temperature of LP turbine steam in Enthalpy of LP turbine steam in Entropy of LP turbine steam in

169.72 kg/s 808 K 3433.78 kJ/kg 6.56 kJ/kg K 147.22 kg/s 808 K 3541.36 kJ/kg 7.44 kJ/kg K 124.53 kg/s 438.4 K 2804.62 kJ/kg 7.55 kJ/kg K

Mass flow rate of HP turbine steam extraction 1 m4Temperature of HP turbine steam extraction 1 T4 Enthalpy of HP turbine steam extraction 1 h4 Entropy of HP turbine steam extraction 1 S4 Mass flow rate of HP turbine steam extraction 2 m5 Temperature of HP turbine steam extraction 2 T5 Enthalpy of HP turbine steam extraction 2 h5 Entropy of HP turbine steam extraction 2 **S5** Mass flow rate of HP turbine steam out m6 Temperature of HP turbine steam out T6 Enthalpy of HP turbine steam out h6 Entropy of HP turbine steam out **S6**

9.28 kg/s 634.1 K 3118.99 kJ/kg 6.06 kJ/kg K 13.22 kg/s 582.2 K 3027.73 kJ/kg 6.65 kJ/kg K 147.22 kg/s 583 K 3027.73 kJ/kg 6.63 kJ/kg K

Mass flow rate of IP turbine steam extraction 1 m7Temperature of IP turbine steam extraction 1 T7 Enthalpy of IP turbine steam extraction 1 h7 Entropy of IP turbine steam extraction 1 **S7** Mass flow rate of IP turbine steam extraction 2 m8 Temperature of IP turbine steam extraction 2 **T8** Enthalpy of IP turbine steam extraction 2 h8 Entropy of IP turbine steam extraction 2 **S8** Mass flow rate of IP turbine steam extraction 3 m9 Temperature of IP turbine steam extraction 3 Т9 Enthalpy of IP turbine steam extraction 3 h9 Entropy of IP turbine steam extraction 3 **S9**

4.50 kg/s 710.4 K 3340.43 kJ/kg 7.49 kJ/kg K 6.69 kg/s 623.3 K 3165.03 kJ/kg 7.50 kJ/kg K 6.47 kg/s 531.4 K 2983.78 kJ/kg 7.52 kJ/kg K

Mass flow rate of IP turbine steam extraction 4 m10 Temperature of IP turbine steam extraction 4 516 K T10 Enthalpy of IP turbine steam extraction 4 h10 Entropy of IP turbine steam extraction 4 S10 Mass flow rate of IP turbine steam out m11 Temperature of IP turbine steam out T11 Enthalpy of IP turbine steam out h11 Entropy of IP turbine steam out S11 Mass flow rate of LP turbine steam extraction 1 m12 Temperature of LP turbine steam extraction 1 T12 Enthalpy of LP turbine steam extraction 1 h12 Entropy of LP turbine steam extraction 1 S12 Mass flow rate of LP turbine steam out m13 Temperature of LP turbine steam out T13 Enthalpy of LP turbine steam out h13 Entropy of LP turbine steam out र1२

5.03 kg/s 2959.08 kJ/kg 7.88 kJ/kg K 124.53 kg/s 438.4 K 2804.62 kJ/kg 7.55 kJ/kg K 5.36 kg/s 357.9 K 2652.25 kJ/kg 7.62 kJ/kg K 119177 kg/s 319.9 K 2437.09 kJ/kg 7 609 kI/ko K

Energy Analysis of Turbine



Exergy Analysis of Turbine



Turbine Analysis Result

Energy Analysis Result



Conclusion

- Coal Analysis result gives good quality coal generate less ash and its calorific value is high. Best quality coal which is use in Wanakbori Thermal Power Plant is imported coal.
- Boiler performance evaluated using direct and indirect method. Both methods gives similar efficiency.
- Based on indirect method it is found that loss due to flue gas and moisture is higher.
- Energy efficiency of component is in most of cases is more than Exergy efficiency of the component of Thermal Power Plant.
- By Finding Energy Efficiency and Exergy Efficiency component wise we get component performance.
- It is found that all component performance is good.

Future work

- To do component wise exergy and energy analysis of remaining components of boiler.
- To find exergy and energy analysis at various load of Thermal Power Plant.

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THANK YOU